



Micronutrient constraints in sodic soils of Israna, Haryana (India)

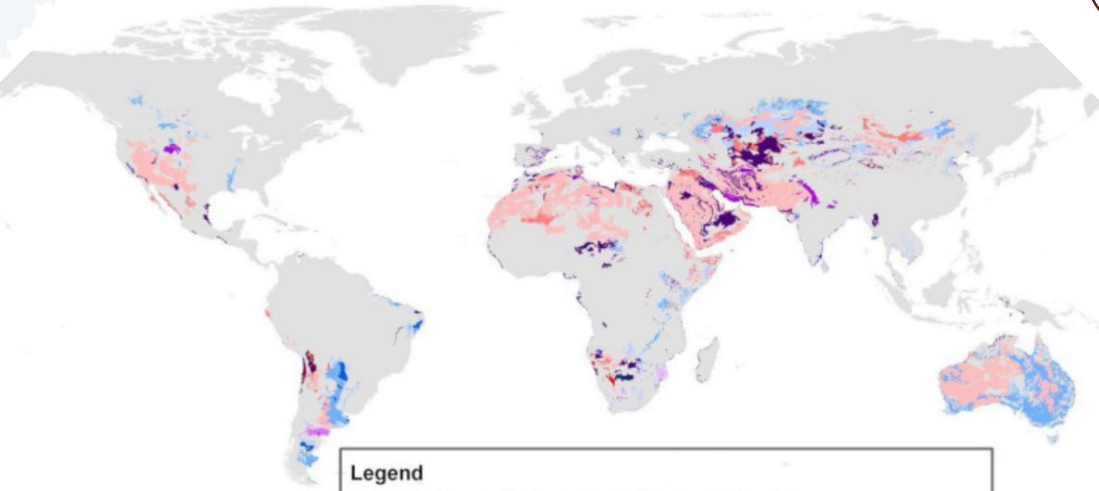
Dr. Seema Chahal, JNU, India



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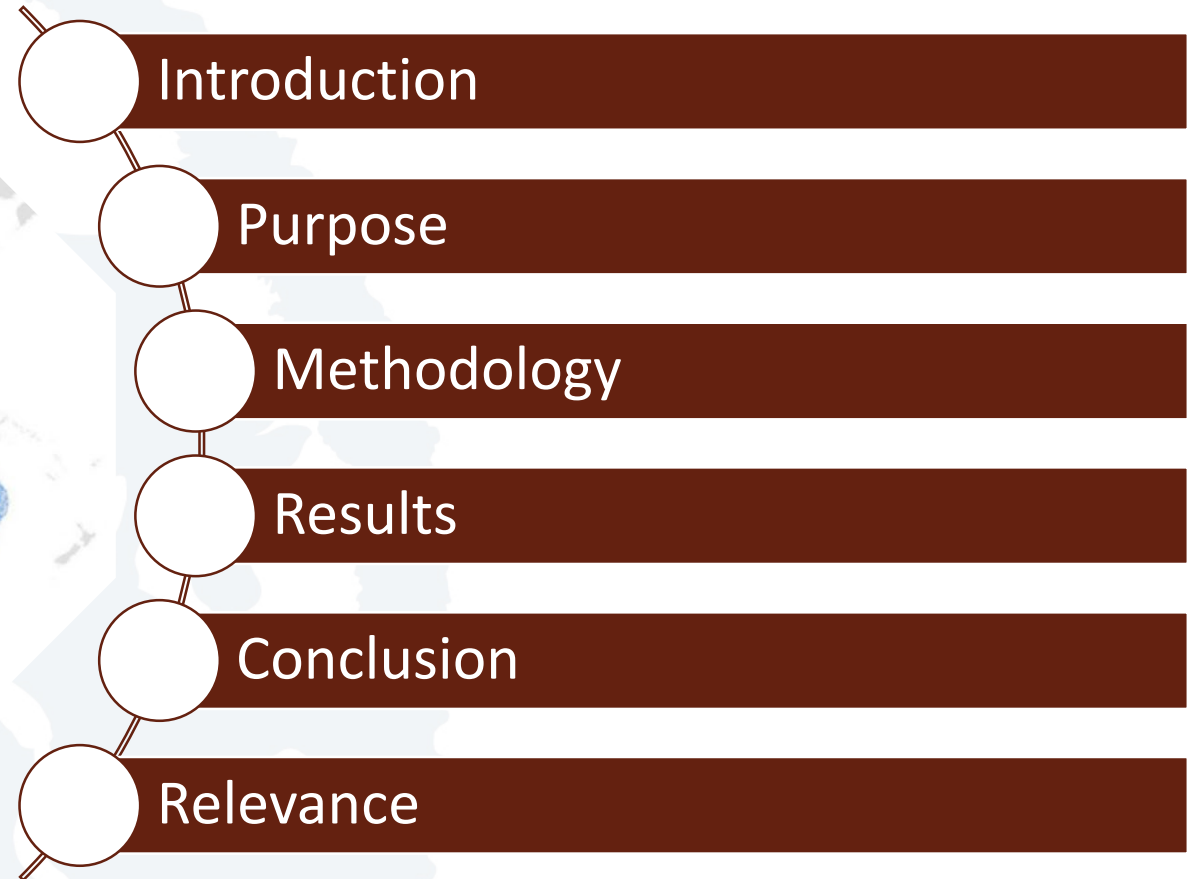
Outline of the presentation



Legend
Type and severity levels of salt-affected soils

saline slight	sodic slight	saline-sodic slight
saline moderate	sodic moderate	saline-sodic moderate
saline high	sodic high	saline-sodic high
saline extreme	sodic extreme	saline-sodic extreme

Wicke et al, 2011



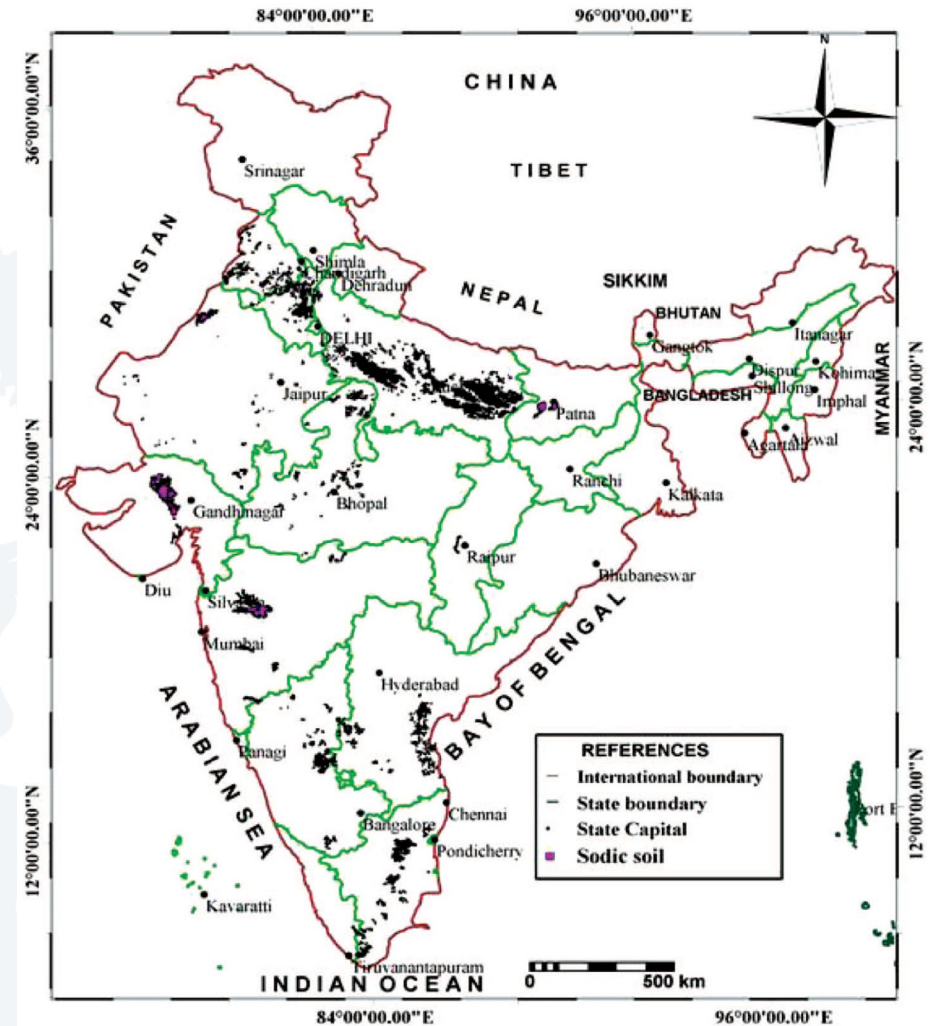
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Sodic soils

Some statistics...

- Sodic soils are widespread in **semi-arid** subtropical regions of world (*Singh et al., 2013*)
- **Global extent : 581 Mha** (*CSSRI-Karnal, 2015*)
- **Largest sodic area: Asia and Australia** (*Leogrande and Vitti, 2018*)
- **Asian continent: 249 Mha** (*Singh et al., 2013*)
- **India: 3,770,659 ha** (*Chhabra et al., 2005*)
- **Haryana: 2,55,700 ha**
- **Mostly in its central part** (*Chhabra et al., 2005*)



Mandal et al., 2009

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GLOBAL SOIL
PARTNERSHIP

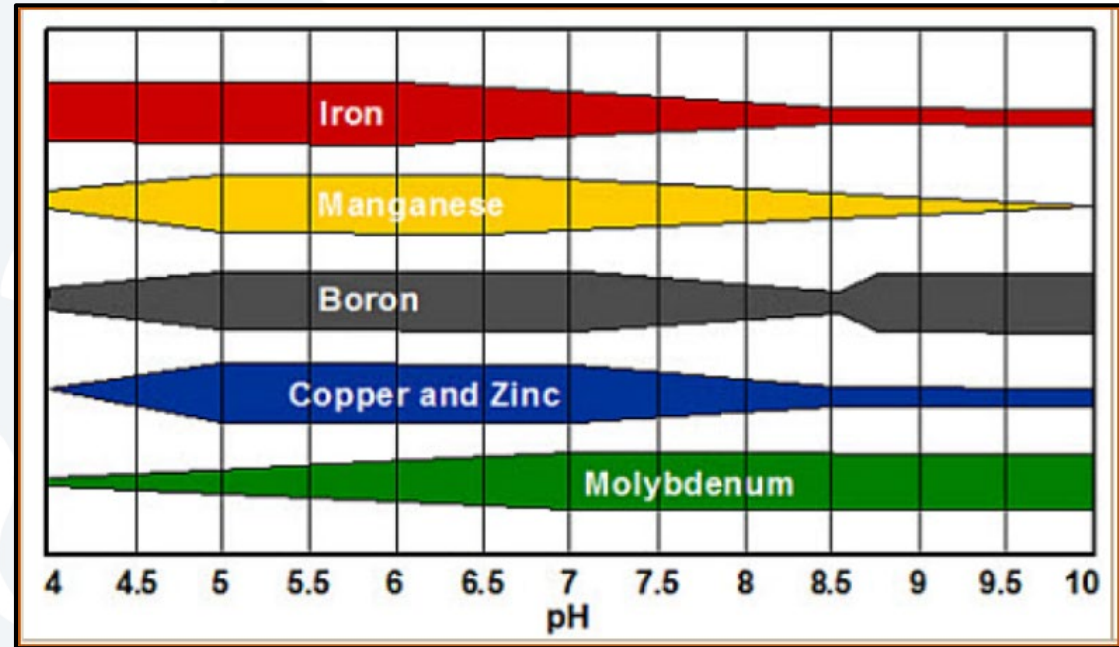
Micronutrient (MN) constraints in sodic soils

- Low in plant available **cationic MNs**
- **Anionic MNs** often display **toxicity**
- Various edafic factors affect their availability in soil (*Qadir et al. 2006*)
- MN deficiency is **a global problem**
- Affects 2 billion people

(*De-Regil et al. 2013; Voortman and Bindraban 2015; Dhaliwal et al. 2019*)

- Reduces crop output
- Adversely impacts human health

(*Graham and Welch 2000*)



<https://www.mssoy.org/uploads/files/micronutrients-pioneer-mar-2017.pdf>

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Purpose of the investigation

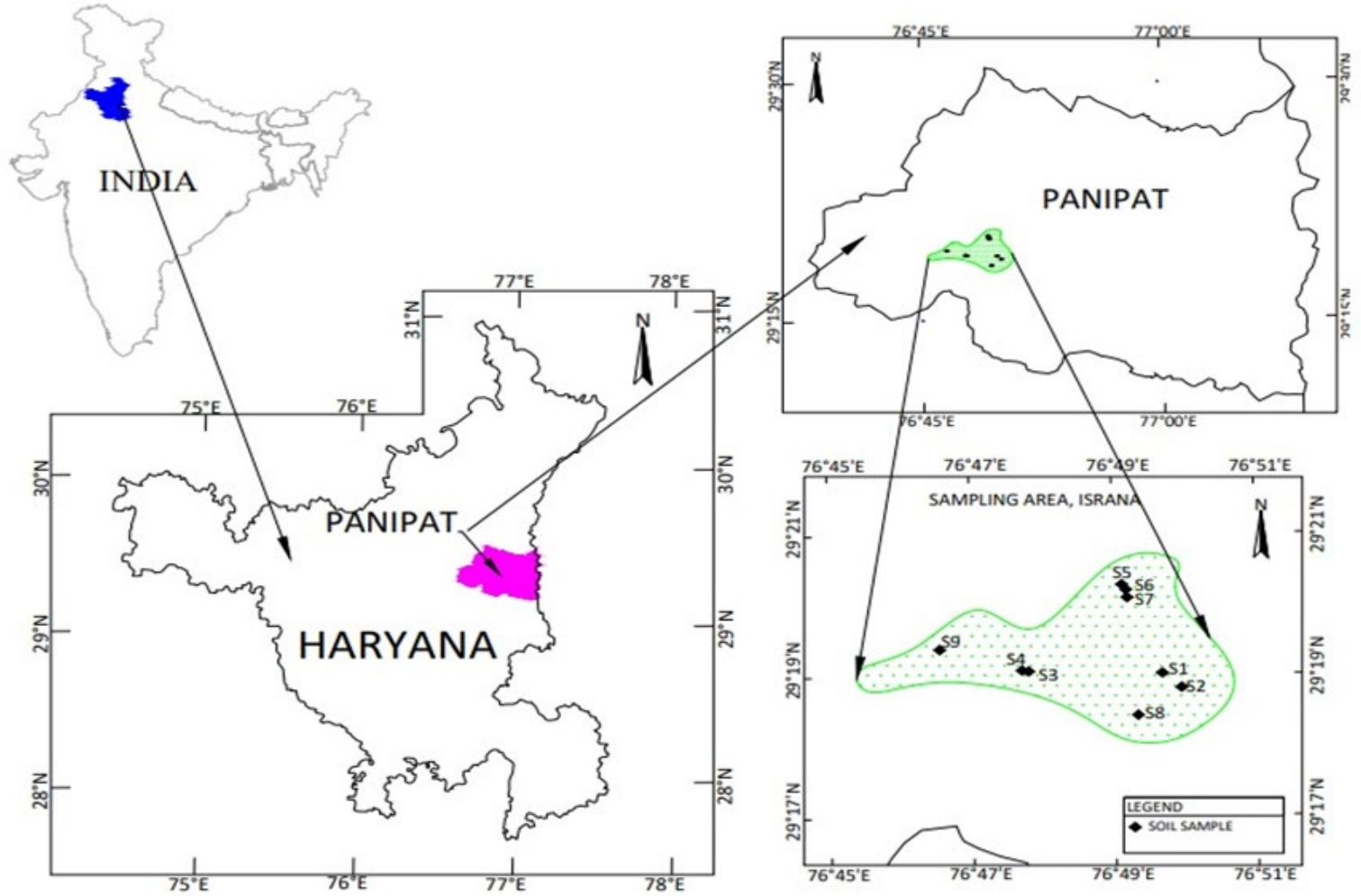
- To determine the availability of B, Cu, Fe, Mn, Mo and Zn in sodic and farmland soils of Israna, Panipat
- To study the influence of various edafic variables on their phyto-availability

Methodology

Soil sampling and
processing

Soil analyses

Data interpretation

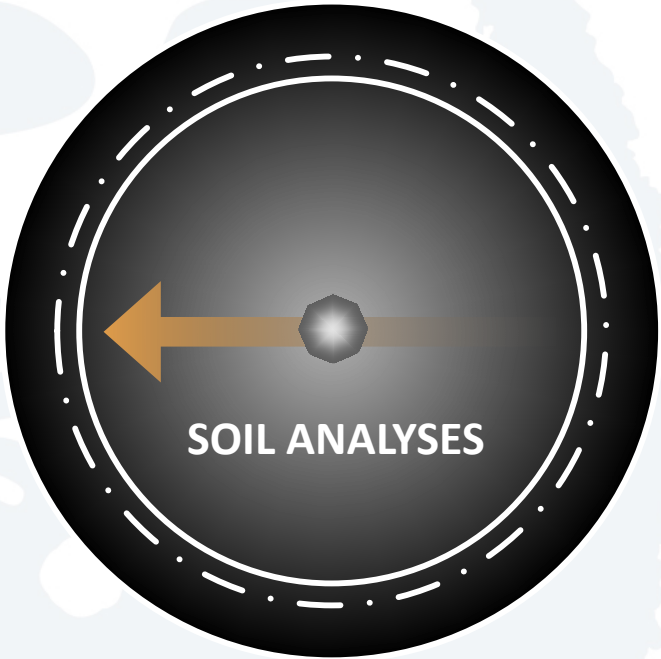


Exchangeable cations
Ammonium acetate method (Richards 1954)
Cation exchange capacity (CEC)
Ammonium acetate method (Bashour and Sayegh, 2007)

Total calcium carbonate content
Titrimetric method (Bashour and Sayegh, 2007)

Available B
Hot water soluble method (hws-B) (MoA, 2011)
Available Mo
Ammonium acetate method (MoA, 2011)

pH and EC
Eutech Cyberscan 510
Texture
Laser particle size analyser Microtrac S3500 (Konert and Vandenberghe, 1997)

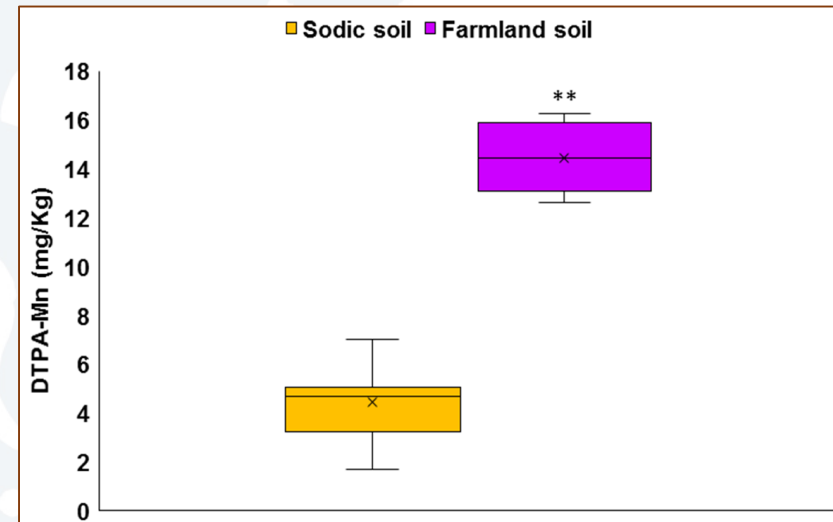
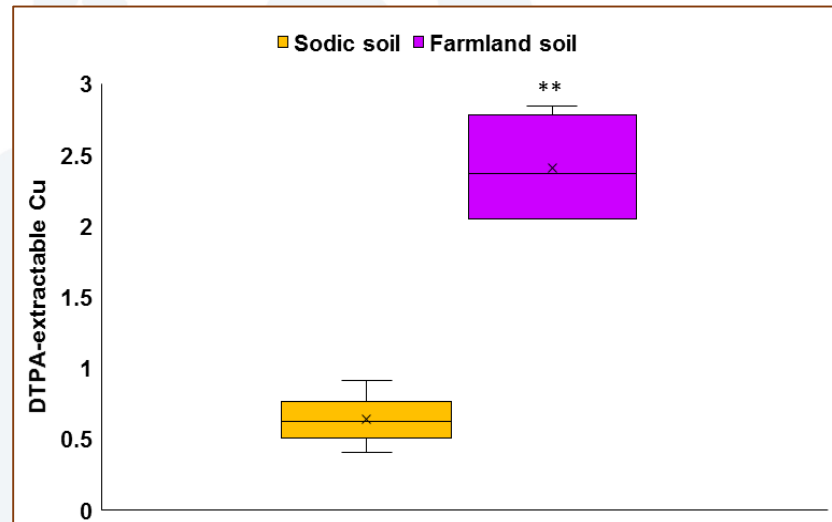
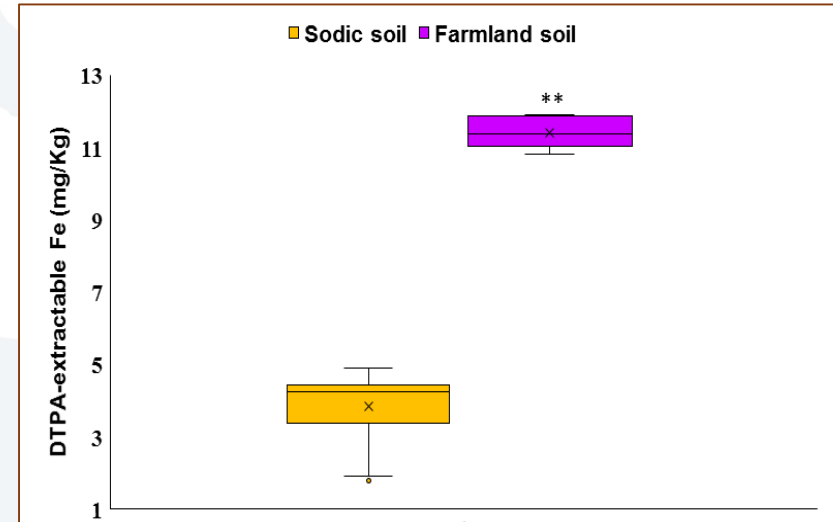
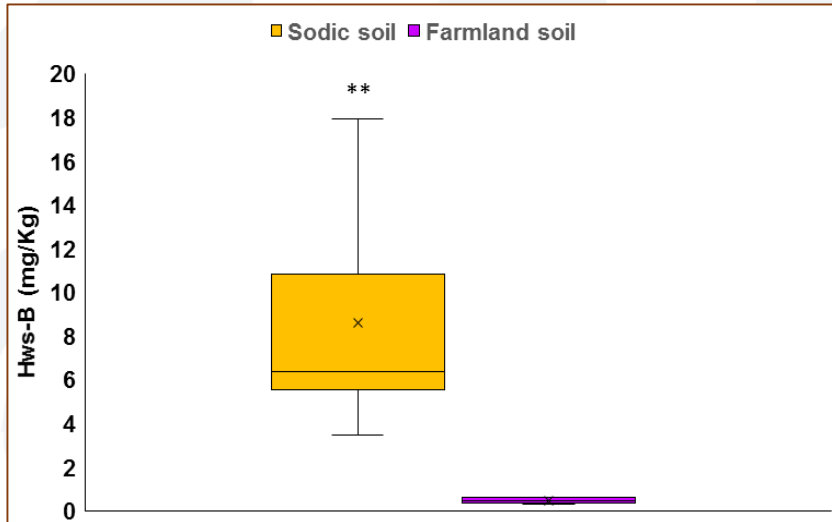


Available Cu, Fe, Mn & Zn
DTPA extraction method (Lindsay and Norvell, 1978)

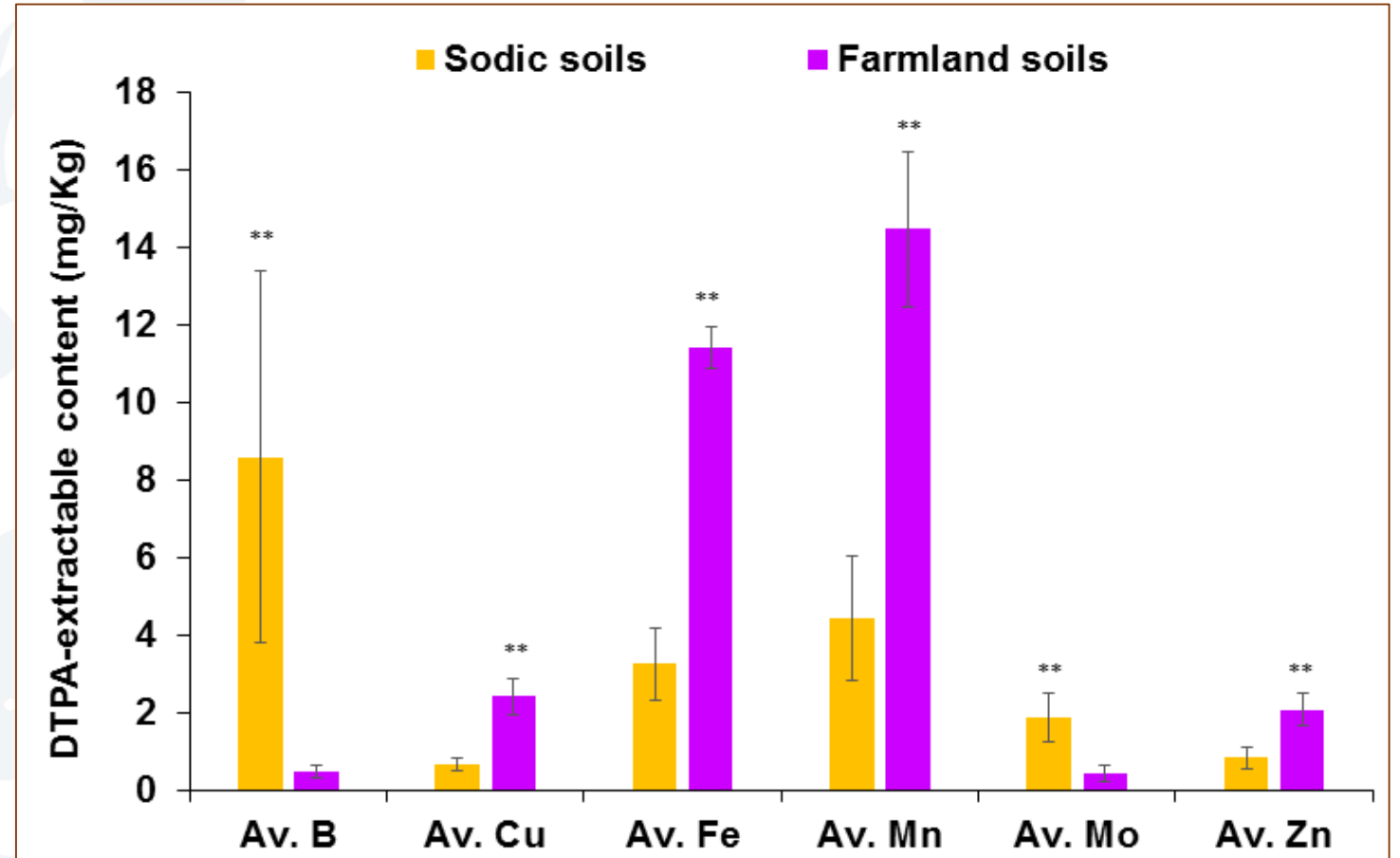
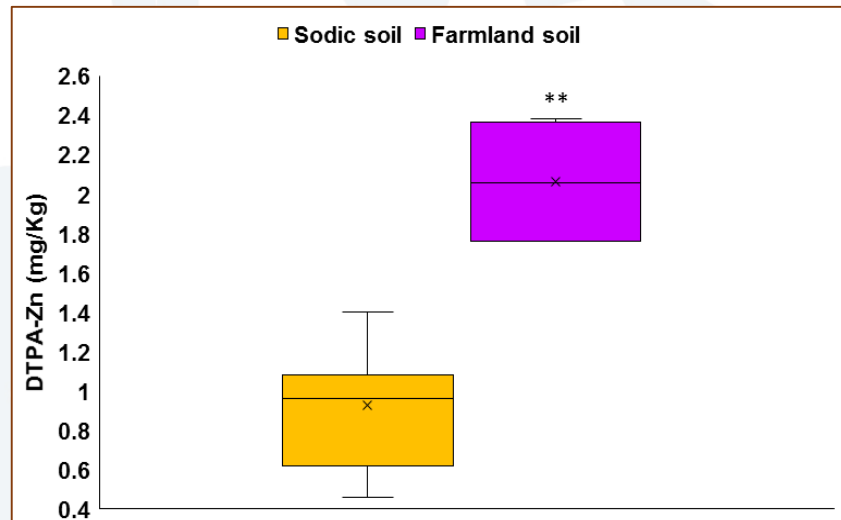
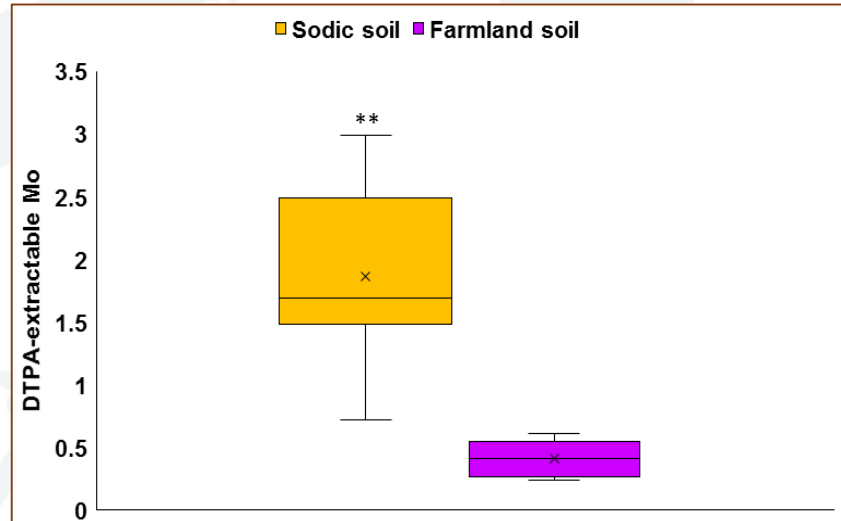
Soil properties and MN availability

Soil properties	Soil Types		Available micronutrient concentration (mg/Kg)	Soil Types	
	Sodic soils	Farmland soils		Sodic soils	Farmland soils
pH	9.41	7.39	B	8.59	0.46
EC (dS/m)	1.38	2.42	Cu	0.65	2.41
ESP (%)	57.64	7.97	Fe	3.84	11.4
Calcite (%)	21.38	11	Mn	4.44	14.46
Sand (%)	15.96	18.59	Mo	1.87	0.41
Silt (%)	69.04	66.44	Zn	0.82	2.06
Clay (%)	14.99	14.98			
CEC (cmol _e /Kg)	10.06	13.61			

Available content of MNs in surface soils



Available content (contd...)



Correlation and regression analyses

	Av. B	Av. Cu	Av. Fe	Av. Mn	Av. Mo	Av. Zn	pH	EC	ESP	Calcite	Sand	Silt	Clay	
Av. B	1	-0.83	-0.83	-0.69	0.29	-0.43	0.86	-0.5	0.29	0.61	-0.02	0.13	-0.13	1.0
Av. Cu	-0.83	1	0.7	0.7	-0.09	0.54	-0.81	0.53	-0.47	-0.55	0.08	-0.09	-0.04	0.8
Av. Fe	-0.83	0.7	1	0.74	-0.12	0.26	-0.78	0.82	0.15	-0.4	-0.24	0.05	0.38	0.7
Av. Mn	-0.69	0.7	0.74	1	0.32	0.54	-0.87	0.78	-0.05	0.02	-0.27	0.39	0.002	0.5
Av. Mo	0.29	-0.09	-0.12	0.32	1	-0.004	-0.04	0.23	-0.03	0.5	-0.61	0.75	0.15	0.3
Av. Zn	-0.43	0.54	0.26	0.54	-0.004	1	-0.67	0.1	-0.46	0.07	-0.07	0.37	-0.36	0.2
pH	0.86	-0.81	-0.78	-0.87	-0.04	-0.67	1	-0.61	0.25	0.27	0.24	-0.29	-0.06	0.0
EC	-0.5	0.53	0.82	0.78	0.23	0.1	-0.61	1	0.34	-0.03	-0.32	0.22	0.3	-0.2
ESP	0.29	-0.47	0.15	-0.05	-0.03	-0.46	0.25	0.34	1	0.47	-0.07	-0.01	0.14	-0.3
Calcite	0.61	-0.55	-0.4	0.02	0.5	0.07	0.27	-0.03	0.47	1	-0.16	0.52	-0.38	-0.5
Sand	-0.02	0.08	-0.24	-0.27	-0.61	-0.07	0.24	-0.32	-0.07	-0.16	1	-0.85	-0.74	-0.7
Silt	0.13	-0.09	0.05	0.39	0.75	0.37	-0.29	0.22	-0.01	0.52	-0.85	1	0.28	-0.8
Clay	-0.13	-0.04	0.38	0.002	0.15	-0.36	-0.06	0.3	0.14	-0.38	-0.74	0.28	1	-1.0

Metal	Extraction method	Equations	R ²
B	HWS	Y = 6.15 pH - 49.30	0.75**
		Y = 5.37 pH + 0.45 Calcite - 51.65	0.90**
		Y = 6.16 pH + 0.28 Calcite - 0.34 Silt - 78.84	0.94*
Cu	DTPA	Y = 2.41 - 0.19 pH	0.65**
		Y = 2.48 - 0.17 pH - 0.01 Calcite	0.77**
		Y = 2.75 - 0.16 pH - 0.02 Calcite - 0.02 Clay	0.82*
		Y = 2.27 - 0.11 pH - 0.02 Calcite - 0.02 Clay + 0.13 EC	0.88*
Fe	DTPA	Y = 2.21 - 0.07 pH - 0.02 Calcite - 0.02 Clay + 0.21 EC - 0.01 ESP	0.93**
		Y = 1.20 + 1.91 EC	0.67**
		Y = 2.96 + 1.88 EC - 0.08 Calcite	0.81**
Mn	DTPA	Y = 7.58 + 1.42 EC - 0.06 Calcite - 0.46 pH	0.87*
		Y = 24.02 - 2.08 pH	0.76**
		Y = 16.40 - 1.50 pH + 1.58 EC	0.86**
Mo	Ammonium acetate (pH = 7)	Y = 16.87 - 1.71 pH + 1.40 EC + 0.08 Calcite	0.90*
Zn	DTPA	Y = 0.14 Silt - 8.09	0.57**
		Y = 3.52 - 0.29 pH	0.44**
		Y = 4.28 - 0.30 pH - 0.04 Clay	0.60*
		Y = 2.10 - 0.26 pH - 0.05 Clay + 0.03 Silt	0.70*

Conclusion

- The micronutrients followed the order:
 - **Sodic soils:** B > Mn > Fe > Mo > Zn > Cu
 - **Farmland soils:** Mn > Fe > Cu > Zn > B > Mo
- Sodic soils had *significantly lower concentrations* of Cu, Fe, Mn and Zn than non-sodic farmland soils
- Mo and B were found to be in *toxic range* in sodic soils
- **pH** and **EC** exhibited strong influences on MN availability
- Regression models suggested that the phyto-availability of MNs was largely controlled by pH, followed by other soil attributes
- **Judicious management** of these nutrients requires regular monitoring of
 - a) their status
 - b) various soil variables

Relevance

- Sodic soils *cover huge expanse* of terrestrial surface
- Their global extent is projected to *increase in future* as a result of *climate change induced variations*
- **Socio-ecological significance** of such degraded lands *will grow further* in coming years
- Sodic soils display a range of nutrient imbalances
- Affect the crop quality and productivity adversely
- *Profoundly influence* human health as soil is an important source of nutrients in our food supply

You can trace every sickness, every disease and every ailment to a mineral deficiency

Dr. Linus Pauling, Nobel laureate (1954)



Thank you !

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