

# MALTOL

(TENTATIVE)

*Revised specifications prepared at the 65<sup>th</sup> JECFA (2005) and published in FNP 52 Add13 (2005), superseding specifications prepared at the 25<sup>th</sup> JECFA (1981) and published in FNP 52 (1992). An ADI of 0-1 mg/kg bw established at 25<sup>th</sup> JECFA (1981)*

*Information on functional uses and method of assay required*

## SYNONYMS

INS No. 636

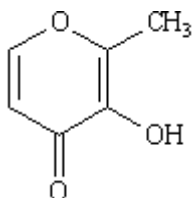
## DEFINITION

Chemical names 3-Hydroxy-2-methyl-4-pyrone

C.A.S. number 118-71-8

Chemical formula  $C_6H_6O_3$

Structural formula



Formula weight 126.11

Assay Not less than 99%

## DESCRIPTION

White to off-white crystalline powder having a characteristic caramel-butterscotch odour

## FUNCTIONAL USES

Flavour enhancer, stabilizer, flavouring agent (see Flavouring agents monograph No. 1480)

## CHARACTERISTICS

### IDENTIFICATION

Solubility (Vol.4) Soluble in water and ethanol

Melting range (Vol. 4) 160 - 164°

Test for phenol Dissolve 0.1 g of the sample in 10 ml of ethanol and add 3 drops of ferric chloride TS. A reddish violet colour is produced.

Precipitation test Dissolve 0.5 g of the sample in 10 ml of sodium hydroxide TS and pass carbon dioxide through the solution. White crystals are formed; collect and recrystallize from dilute ethanol. The crystals melt between 160 - 164°.

Iodoform reaction Dissolve 0.1 g of the sample in 5 ml dioxane, add 1 ml of sodium hydroxide TS, and add sufficiently iodine-potassium iodide TS (Iodine TS) with shaking until the colour remains. Heat on a water bath for 5 min. Yellow crystals are formed.

## PURITY

Lead (Vol. 4)

Not more than 1 mg/kg

Determine using an atomic absorption technique appropriate to the specified level. The selection of sample size and method of sample preparation may be based on the principles of the methods described in Volume 4, "Instrumental Methods".

## **METHOD OF ASSAY**

### Standard Solution

Transfer about 50 mg of Maltol Reference Standard (available from the United States Pharmacopeia, 12601 Twinbrook Parkway, Rockville, Md. 20852, USA), accurately weighed, into a 250-ml flask, dilute to volume with 0.1 N hydrochloric acid, and mix. Pipet 5 ml of this solution into a 100-ml volumetric flask, dilute to volume with 0.1 N hydrochloric acid, and mix.

### Assay Solution

Transfer about 50 mg of the sample, accurately weighed, into a 250-ml flask, dilute to volume with 0.1 N hydrochloric acid. Pipet 5 ml of this solution into a 100-ml volumetric flask, dilute to volume with 0.1 N hydrochloric acid, and mix.

### Procedure

Determine the absorbance of each solution in a 1-cm quartz cell at 274 nm using 0.1 N hydrochloric acid as the blank.

Calculate the percent of Maltol in the sample by the formula:

$$\% \text{ of Maltol} = 100 \times W_S \times A_A / A_S \times W_A$$

where

$A_A$  = absorbance of the Assay Solution

$A_S$  = absorbance of the Reference Standard Solution

$W_A$  = weight in mg of the Assay solution (sample)

$W_S$  = weight in mg of the Reference Standard