

# CALCIUM DL-MALATE

Prepared at the 67<sup>th</sup> JECFA (2006), published in FAO JECFA Monographs 3 (2006), superseding specifications prepared at the 27<sup>th</sup> JECFA (1983) and published in FNP 52 (1992) and in the Combined Compendium of Food Additive Specifications, FAO JECFA Monographs 1 (2005). Metals and arsenic specifications were revised at the 59<sup>th</sup> JECFA (2002). A group ADI 'not specified' for malic acid and its Ca, K and Na salts was established at the 23<sup>rd</sup> JECFA (1979).

## SYNONYMS

DL-Monocalcium malate; INS No. 352(ii)

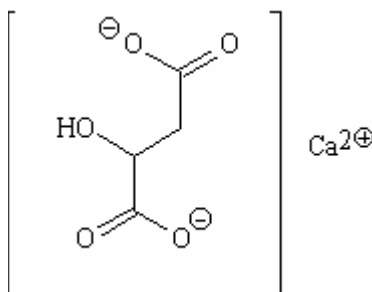
## DEFINITION

Chemical names Monocalcium DL-malate, 2-hydroxybutanedioic acid monocalcium salt

C.A.S. number 17482-42-7

Chemical formula  $C_4H_4CaO_5$

Structural formula



Formula weight 172.1

Assay Not less than 97.5% after drying

## DESCRIPTION

White, colourless powder

## FUNCTIONAL USES

Buffering agent

## CHARACTERISTICS

### IDENTIFICATION

Solubility (Vol. 4) Slightly soluble in water, insoluble in ethanol

Test for malate (Vol. 4) Passes test  
Test 100 ml of a saturated solution of the sample

Test for calcium (Vol. 4) Passes test

## PURITY

<u>Loss on drying</u> (Vol. 4)	Not more than 2% (110°, 3 h)
<u>Fluoride</u> (Vol. 4)	Not more than 30 mg/kg (Method III)
<u>Fumaric acid and maleic acid</u> (Vol. 4)	Not more than 1.0% of fumaric acid and not more than 0.05% of maleic acid
<u>Lead</u> (Vol.4)	Not more than 2 mg/kg Determine using an AAS/ICP-AES technique appropriate to the specified level. The selection of sample size and method of sample preparation may be based on the principles of the methods described in Volume 4.

**METHOD OF ASSAY** Weigh accurately about 0.4 g of the sample, previously dried at 110° for 3 h, dissolve in a mixture of 10 ml of water and 2 ml of dilute hydrochloric acid TS, and dilute to about 100 ml with water. While stirring (preferably with a magnetic stirrer) add about 30 ml of 0.05 M disodium ethylenediaminetetraacetate from a 50-ml buret, then add 15 ml of sodium hydroxide TS and 300 mg of hydroxynaphtol blue indicator, and continue the titration to a blue endpoint. Each ml of 0.05 M disodium ethylenediaminetetraacetate is equivalent to 8.607 mg of C<sub>4</sub>H<sub>4</sub>CaO<sub>5</sub>.